Intermolecular Forces And Liquids And Solids

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vapor pressure purdue university Oct 03 2022 although dipole dipole forces and london dispersion forces also exist between ethyl alcohol molecules the strong hydrogen bonding interactions are responsible for the much lower vapor pressure compared to ethyl ether temperature at a higher temperature more molecules have enough energy to escape from the liquid or solid at a lower

surface tension purdue university Nov 04 2022 surface tension surface tension is measured as the energy required to increase the surface area of a liquid by a unit of area the surface tension of a liquid results from an imbalance of intermolecular attractive forces the cohesive forces between molecules a molecule in the bulk liquid experiences cohesive forces with other molecules in all directions